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## Structure Reports

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## Key indicators

Single-crystal X-ray study

## $T=110 \mathrm{~K}$

Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.033$
$w R$ factor $=0.084$
Data-to-parameter ratio $=22.8$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## Tricarbonyl $\left\{\eta^{6}\right.$-(S,S)-1-[2-methoxy-3,6-bis(tri-methylsilyl)phenyl]-N,N-dimethylethylamine\}chromium(0)

The title complex, $\left[\mathrm{Cr}\left(\mathrm{C}_{17} \mathrm{H}_{33} \mathrm{NOSi}_{2}\right)(\mathrm{CO})_{3}\right]$, has the typical three-legged piano-stool structure expected for $\eta^{6}$-arenetricarbonylchromium compounds. The conformation is staggered with respect to the methoxy group of the arene ring and one of the carbonyl ligands.

## Comment

In the course of our work (Totev et al., 2004; Braun et al., 2004; Englert et al., 2004; Salzer, 2003) on the diastereoselective synthesis of planar chiral arenetricarbonylchromium complexes, we synthesized and structurally characterized the title complex, (I) (Fig. 1). The complex crystallizes in the noncentrosymmetric orthorhombic space group $P 2_{1} 2_{1} 2_{1}$.

(I)

The non-H atoms attached to the complexed arene ring are displaced from the least-squares plane defined by the atoms in the aromatic ring (C104-C109) to various extents; a considerable distortion of -0.193 (1) $\AA$ (towards the Cr atom) is observed for Si1, whereas Si 2 shows a displacement of only


View of (I), with $30 \%$ displacement ellipsoids and H atoms shown as spheres of arbitrary radii.

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0.068 (1) $\AA$ (away from the Cr atom). Atom C 112 is displaced by 0.088 (2) $\AA$ and O1 shows the smallest displacement of -0.006 (1) $\AA$. The $\mathrm{N}-\mathrm{C} 112-\mathrm{C} 108-\mathrm{C} 107$ torsion angle is $133.5(2)^{\circ}$. The lone pair of electrons of the N atom of the chiral side chain point towards the adjacent trimethylsilyl group $[\mathrm{N} \cdots \operatorname{si2}=3.059$ (3) $\AA$ ], while the two $N$-methyl groups are pointing away from Si 2 to minimize steric hindrance.

## Experimental

Compound (I) was synthesized by thermal complexation of $(S, S)$ -1-[2-methoxy-3,6-bis(trimethylsilyl)phenyl]- $N, N$-dimethylethylamine using Kündig's reagent (naphthalene) $\mathrm{Cr}(\mathrm{CO})_{3}$ (Kündig et al., 1985) in a high-pressure Schlenk tube in an $n-\mathrm{Bu}_{2} \mathrm{O} /$ tetrahydrofuran mixture. After work-up, yellow single crystals were obtained by slow diffusion of hexanes into a diethyl ether solution at 243 K .

## Crystal data

$\left[\mathrm{Cr}\left(\mathrm{C}_{17} \mathrm{H}_{33} \mathrm{NOSi}_{2}\right)(\mathrm{CO})_{3}\right]$
$M_{r}=459.65$
Orthorhombic, $P_{2} 2_{1} 2_{1} 2_{1}$
$a=7.1361$ (12) £
$b=17.251$ (3) $\AA$
$c=19.541$ (3) $\AA$
$V=2405.7$ (7) $\AA^{3}$
$Z=4$
$D_{x}=1.269 \mathrm{Mg} \mathrm{m}^{-3}$

## Data collection

Bruker SMART APEX CCD diffractometer
$\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.779, T_{\text {max }}=0.982$
33064 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.033$
$w R\left(F^{2}\right)=0.084$
$S=0.92$
5984 reflections
263 parameters
H -atom parameters constrained

## Mo $K \alpha$ radiation

Cell parameters from 8096 reflections
$\theta=1.6-28.3^{\circ}$
$\mu=0.60 \mathrm{~mm}^{-1}$
$T=110$ (2) K
Rod, yellow
$0.44 \times 0.03 \times 0.03 \mathrm{~mm}$

5984 independent reflections
5037 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.071$
$\theta_{\text {max }}=28.3^{\circ}$
$h=-9 \rightarrow 9$
$k=-22 \rightarrow 22$
$l=-26 \rightarrow 26$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0411 P)^{2}\right] \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.38 \mathrm{e}^{-3} \\
& \Delta \rho_{\min }=-0.28 \mathrm{e} \AA^{-3} \\
& \text { Absolute structure: Flack (1983), } \\
& 2578 \text { Friedel pairs } \\
& \text { Flack parameter }=-0.006(18)
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\mathrm{A},{ }^{\circ}$ ).

| $\mathrm{Cr} 1-\mathrm{C} 3$ | $1.844(3)$ | $\mathrm{Cr} 1-\mathrm{C} 106$ | $2.236(2)$ |
| :--- | ---: | :--- | :--- |
| $\mathrm{Cr} 1-\mathrm{C} 2$ | $1.846(3)$ | $\mathrm{Cr} 1-\mathrm{C} 104$ | $2.246(2)$ |
| $\mathrm{Cr} 1-\mathrm{C} 1$ | $1.850(3)$ | $\mathrm{Cr} 1-\mathrm{C} 107$ | $2.270(2)$ |
| $\mathrm{Cr} 1-\mathrm{C} 105$ | $2.214(3)$ | $\mathrm{Cr} 1-\mathrm{C} 108$ | $2.272(2)$ |
| $\mathrm{Cr} 1-\mathrm{C} 109$ | $2.235(2)$ |  |  |
| $\mathrm{C} 3-\mathrm{Cr} 1-\mathrm{C} 2$ | $88.76(12)$ | $\mathrm{C} 2-\mathrm{Cr} 1-\mathrm{C} 1$ | $87.98(12)$ |
| $\mathrm{C} 3-\mathrm{Cr} 1-\mathrm{C} 1$ | $85.17(12)$ |  |  |

H atoms were placed in calculated positions, with $\mathrm{C}-\mathrm{H}=0.98 \AA$, and refined as riding, with $U_{\text {iso }}(\mathrm{H})=1.3 U_{\text {eq }}(\mathrm{C})$. Methyl groups were allowed to rotate as rigid groups.

Data collection: SMART (Bruker, 2001); cell refinement: SAINTPlus (Bruker, 1999); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97.

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## References

Braun, W., Calmuschi, B., Haberland, J., Hummel, W., Liese, A., Nickel, T., Stelzer, O. \& Salzer, A. (2004). Eur. J. Inorg. Chem. 11, 2235-2243.
Bruker (1999). SAINT-Plus. Version 6.02. Bruker AXS Inc., Madison, Wisconsin, USA.
Bruker (2001). SMART. Version 5.624. Bruker AXS Inc., Madison, Wisconson, USA
Englert, U., Hu, C., Salzer, A. \& Alberico, E. (2004). Organometallics, 23, 5419-5431.
Flack, H. D. (1983). Acta Cryst. A39, 876-881.
Kündig, E. P., Perret, C., Sprichinger, S. \& Bernardinelli, G. (1985). J. Organomet. Chem. 286, 183-200.
Salzer, A. (2003). Coord. Chem. Rev. 242, 59-72.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Spek, A. L. (2003). J. Appl. Cryst, 36, 7-13.
Totev, D., Salzer, A., Carmona, D., Oro, L. A., Lahoz, F. J. \& Dabrinovitch, I. T. (2004), Inorg. Chim. Acta, 357, 2889-2898.

